

A stacked pyrazolo[3,4-*d*]pyrimidine-based flexible molecule: the effect of a bulky benzyl group on intermolecular stacking in comparison with methyl and ethyl groups¹

Kamlakar Avasthi,^a Ashish Tewari,^a Diwan S. Rawat,^a
Ashoke Sharon^b and Prakas R. Maulik^{b*}

^aMedicinal Chemistry Division, Central Drug Research Institute, Lucknow 226 001, India, and ^bMolecular and Structural Biology Division, Central Drug Research Institute, Lucknow 226 001, India

Correspondence e-mail: maulik_prakas@yahoo.com

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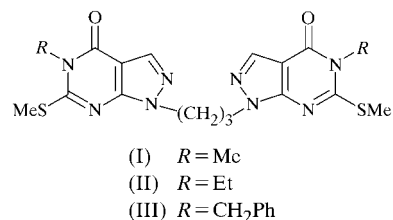
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In the crystal structure of 1,1'-(1,3-propanediyl)bis(5-benzyl-6-methylsulfanyl-4,5-dihydro-1*H*-pyrazolo[3,4-*d*]pyrimidin-4-one), C₂₉H₂₈N₈O₂S₂, the pairs of pyrazolo[3,4-*d*]pyrimidine rings stack as a result of intramolecular π - π interactions between the heterocyclic rings. The folded molecules are further stacked in pairs, due to intermolecular aromatic π - π interactions and C—H...O hydrogen bonds.

Comment

Interactions between aromatic units play a significant role in chemistry (Muller-Dethlefs & Hobza, 2000; Hunter *et al.*, 2001; Tsuzuki *et al.*, 2002), crystal engineering (Desiraju, 1995) and biology. In recent years, we have reported the convenient syntheses (Avasthi *et al.*, 1995, 1998; Avasthi, Rawat *et al.*, 2001) and the X-ray structures (Biswas *et al.*, 1995; Maulik *et al.*, 1998, 2000; Avasthi, Rawat *et al.*, 2001; Avasthi, Aswal & Maulik, 2001) of several novel 'propylene-linker' compounds based on the pyrazolo[3,4-*d*]pyrimidine core, which is isomeric with biologically important purine, as flexible new models for studying aromatic π - π interactions (APPI). Two of these compounds, *viz.* 1,1'-(1,3-propanediyl)bis(5-methyl-6-methylthio-4,5-dihydro-1*H*-pyrazolo[3,4-*d*]pyrimidin-4-one), (I), and 1,1'-(1,3-propanediyl)bis(5-ethyl-6-methylthio-dihydro-1*H*-pyrazolo[3,4-*d*]pyrimidin-4-one), (II), show inter- and intramolecular stacking due to APPI (Maulik *et al.*, 1998; Avasthi, Aswal & Maulik, 2001) when studied using X-ray crystallography. Since the X-ray structures of (I) and (II) are quite similar in having a U-motif for the demonstration of inter- and intramolecular stacking, it was considered worthwhile to replace the *N*-methyl/ethyl group of (I) and (II) with a bulky

N-benzyl group, to determine the robustness of the U-motif and its consequence for the intermolecular stacking from a crystal engineering point of view. In this communication, we report the X-ray structure of 1,1'-(1,3-propanediyl)bis(5-benzyl-6-methylsulfanyl-4,5-dihydro-1*H*-pyrazolo[3,4-*d*]pyrimidin-4-one), (III), the synthesis of which was described previously by Avasthi *et al.* (1998).



The molecular structure and conformation of (III) are shown in Fig. 1. The structure is folded at the centre of the bridge [C10—C11—C12 114.0 (2)°] due to an intramolecular APPI between the pyrazolo[3,4-*d*]pyrimidine rings. For comparison, the folding angles in (I) and (II) are 115.2 (2) and 114.9 (2)°, respectively. In compound (III), as in (I) and (II), the folded pyrazolo[3,4-*d*]pyrimidine rings are positioned in

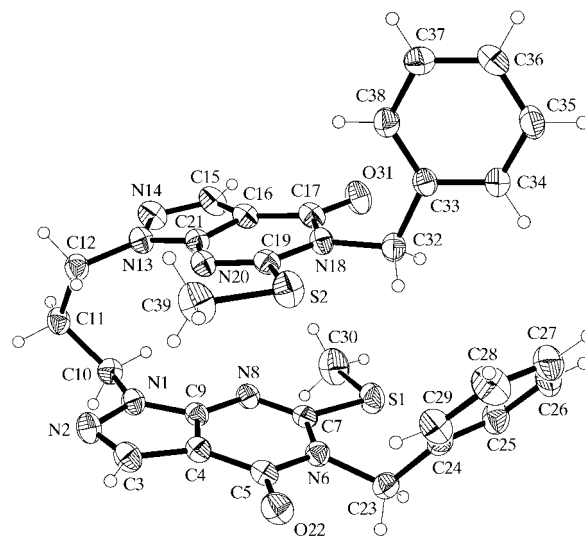


Figure 1

A displacement ellipsoid plot (30% probability), showing the molecular structure of (III) and the atom-labelling scheme.

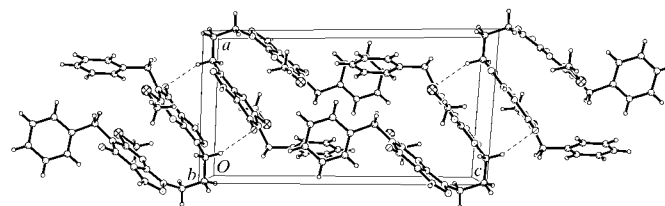


Figure 2

A crystal-packing diagram for (III), showing the intra- and intermolecular π - π stacking between the pyrazolo[3,4-*d*]pyrimidine rings and the intermolecular C—H...O hydrogen bonding (dashed lines).

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such a way that the pyrimidinyl rings overlap only partially. The overlapping six-membered rings are separated by an average distance of 3.428 (3) Å [cf. 3.37 (1) Å in (I) and 3.415 (3) Å in (II)], thus confirming the presence of intramolecular APPI.

The pyrazolo[3,4-*d*]pyrimidine rings in (III) are nearly planar [maximum deviation = −0.048 (2) Å] and the angle between the least-squares planes is 14.5 (1)° [cf. 12.4 (5)° in (I) and 12.5 (1)° in (II)]. The crystal packing (Fig. 2) shows further independent intermolecular stacking between the pyrazolo[3,4-*d*]pyrimidine systems due to π - π interactions. Pairs of pyrazolo[3,4-*d*]pyrimidine rings [related by symmetry code (1 - *x*, 1 - *y*, -*z*)] overlap, with an interplanar separation of 3.370 (2) Å in a 'parallel-displaced' orientation [the dihedral angle of a stacking pair is 1.0 (1)°].

Interestingly, these stacked pyrazolo[3,4-*d*]pyrimidine rings are also connected by intermolecular C-H...O hydrogen bonding (Table 1; Desiraju & Steiner, 1999). Thus, the combination of intra- and intermolecular APPI and intermolecular hydrogen bonding results in the formation of a stacked dimeric unit of (III). The continuous intermolecular stacking present in (I) and (II) is absent in (III). The crystal structure of (III) is stabilized mainly by C-H...O bonding, π - π interactions and van der Waals forces.

Experimental

Compound (III) was synthesized according to the method of Avasthi *et al.* (1998). Diffraction-quality crystals were obtained by slow evaporation of an ethyl acetate solution at room temperature.

Crystal data

C ₂₉ H ₂₈ N ₈ O ₂ S ₂	<i>Z</i> = 2
<i>M_r</i> = 584.71	<i>D_x</i> = 1.379 Mg m ⁻³
Triclinic, <i>P</i> $\bar{1}$	Mo <i>K</i> α radiation
<i>a</i> = 9.150 (1) Å	Cell parameters from 38 reflections
<i>b</i> = 9.491 (1) Å	θ = 4.8–12.5°
<i>c</i> = 16.839 (2) Å	μ = 0.23 mm ⁻¹
α = 83.01 (1)°	<i>T</i> = 293 (2) K
β = 85.26 (1)°	Rectangular, colourless
γ = 76.39 (1)°	0.35 × 0.28 × 0.25 mm
<i>V</i> = 1408.5 (3) Å ³	

Data collection

Bruker P4 diffractometer	<i>h</i> = −1 → 11
$\theta/2\theta$ scans	<i>k</i> = −11 → 11
6623 measured reflections	<i>l</i> = −20 → 20
5527 independent reflections	3 standard reflections
3716 reflections with <i>I</i> > 2 σ (<i>I</i>)	every 97 reflections
<i>R</i> _{int} = 0.021	intensity decay: none
θ_{\max} = 26°	

Table 1

Hydrogen-bonding geometry (Å, °).

<i>D</i> –H... <i>A</i>	<i>D</i> –H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> –H... <i>A</i>
C10–H10A...O22 ⁱ	0.97	2.47	3.367 (3)	154

Symmetry code: (i) 1 - *x*, 1 - *y*, -*z*.

Refinement

Refinement on <i>F</i> ²	$w = 1/[\sigma^2(F_o^2) + (0.0470P)^2 + 0.5548P]$
$R[F^2 > 2\sigma(F^2)] = 0.047$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.121$	$(\Delta/\sigma)_{\max} < 0.001$
<i>S</i> = 1.01	$\Delta\rho_{\max} = 0.18 \text{ e \AA}^{-3}$
5527 reflections	$\Delta\rho_{\min} = -0.28 \text{ e \AA}^{-3}$
372 parameters	
H-atom parameters constrained	

Data collection: XSCANS (Siemens, 1996); cell refinement: XSCANS; data reduction: XSCANS; program(s) used to solve structure: SHELXTL (Bruker, 1997); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: VJ1169). Services for accessing these data are described at the back of the journal.

References

- Avasthi, K., Aswal, S. & Maulik, P. R. (2001). *Acta Cryst.* **C57**, 1324–1325.
- Avasthi, K., Chandra, T. & Bhakuni, D. S. (1995). *Indian J. Chem. Sect. B*, **34**, 944–949.
- Avasthi, K., Rawat, D. S., Chandra, T. & Bhakuni, D. S. (1998). *Indian J. Chem. Sect. B*, **37**, 754–759.
- Avasthi, K., Rawat, D. S., Maulik, P. R., Sarkhel, S., Broder, C. K. & Howard, J. A. K. (2001). *Tetrahedron Lett.* **42**, 7115–7117.
- Biswas, G., Chandra, T., Avasthi, K. & Maulik, P. R. (1995). *Acta Cryst.* **C51**, 2453–2455.
- Bruker (1997). SHELXTL. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.
- Desiraju, G. R. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 2311–2327.
- Desiraju, G. R. & Steiner, T. (1999). *The Weak Hydrogen Bond in Structural Chemistry and Biology*. New York: Oxford University Press Inc.
- Hunter, C. A., Lawson, K. R., Perkins, J. & Urch, C. J. (2001). *J. Chem. Soc. Perkin Trans. 2*, pp. 651–669.
- Maulik, P. R., Avasthi, K., Biswas, G., Biswas, S., Rawat, D. S., Sarkhel, S., Chandra, T. & Bhakuni, D. S. (1998). *Acta Cryst.* **C54**, 275–277.
- Maulik, P. R., Avasthi, K., Sarkhel, S., Chandra, T., Rawat, D. S., Logsdon, B. & Jacobson, R. A. (2000). *Acta Cryst.* **C56**, 1361–1363.
- Muller-Dethlefs, K. & Hobza, P. (2000). *Chem. Rev.* **100**, 143–167.
- Siemens (1996). XSCANS. Version 2.21. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Tsuzuki, S., Honda, K., Uchimaru, T., Mikami, M. & Tanabe, K. (2002). *J. Am. Chem. Soc.* **124**, 104–112.